CHEMISTRY

Chemistry is a science devoted to the manufacture and evaluation of compounds and materials. Chemists have created new types of materials such as Teflon and have synthesized molecules for treating cancer such as derivatives of Taxol. Chemists are also involved in making measurements crucial toward determining the presence of hazardous pollutants in our environment and the safety of our food and water supplies. Chemistry is at the forefront in developing and advancing new technologies to solve problems in agriculture, medicine, electronics, energy and forensics.

A student considering a career in chemistry should have a strong curiosity about the nature of how things work, good problem solving skills and most important of all—a desire to learn. Some background in mathematics and physics is beneficial as these subjects facilitate an understanding of chemistry, which is often called the Central Science because of its importance in understanding both the physical and biological sciences. For this reason alone, chemists are employed in all phases of our economy—industry, government and education (both high school and college).

The Department of Chemistry offers five bachelor’s degrees:

1. Bachelor of Science in Chemistry approved by the American Chemical Society (ACS);
2. Bachelor of Science in Chemistry—Departmental degree that requires less specialization;
3. Bachelor of Science in Chemistry with Secondary Teacher Certification;
4. Bachelor of Science in Chemistry with a Pre-Health/Pre-Law concentration;
5. Bachelor of Science in Medicinal and Biophysical Chemistry

Our undergraduate classes are at the cutting edge and our instructional laboratories are modern and well-equipped with the necessary instrumentation to prepare students for potential careers in fields where a background in chemistry is crucial for success (e.g., medicine, patent law, pharmacy, the environment, nanotechnology, homeland security, public safety).

Courses

CHEM 1014 Chemistry In Civilization (LN)
**Description:** A survey course presenting the concepts and principles of chemistry for students outside the health, science and engineering fields. This course covers the basics of chemistry and chemical contributions to society such as polymers, consumer chemicals, drugs, and radioactivity. May not be used for degree credit with CHEM 1215 or CHEM 1314.

**Credit hours:** 4
**Contact hours:** Lecture: 2 Lab: 2 Contact: 5 Other: 1
**Levels:** Undergraduate
**Schedule types:** Discussion, Lab, Lecture, Combined lecture lab & disc
**Department/School:** Chemistry
**General Education and other Course Attributes:** Scientific Investigation, Natural Sciences

CHEM 1215 Chemical Principles I (LN)
**Prerequisites:** MATH 1483 or MATH 1513 or a higher level math course with a "C" or better or an acceptable math placement score (see placement.okstate.edu) or acceptable AP credit.

**Description:** The beginning chemistry course recommended for students in the applied biological sciences. This course covers chemical principles and their applications to their properties and transformations of matter, including periodic classification of the elements, laws of chemical combination, atomic and molecular structure, and chemical bonding. Course previously offered as CHEM 1015.

**Credit hours:** 5
**Contact hours:** Lecture: 3 Lab: 2 Contact: 6 Other: 1
**Levels:** Undergraduate
**Schedule types:** Discussion, Lab, Lecture, Combined lecture lab & disc
**Department/School:** Chemistry
**General Education and other Course Attributes:** Scientific Investigation, Natural Sciences

CHEM 1225 Chemical Principles II (LN)
**Prerequisites:** CHEM 1215 or CHEM 1314 or CHEM 1414 with a grade of "C" or better; and MATH 1483 or MATH 1513 or higher with a "C" or better or an acceptable math placement score (see placement.okstate.edu); or acceptable AP credit.

**Description:** A continuation of Chemical Principles I for students in the applied biological sciences. Topics include gas laws, chemical equilibria, acid/base chemistry, oxidation/reduction, elementary chemical thermodynamics, and introduction to organic molecules.

**Credit hours:** 5
**Contact hours:** Lecture: 3 Lab: 2 Contact: 6 Other: 1
**Levels:** Undergraduate
**Schedule types:** Discussion, Lab, Lecture, Combined lecture lab & disc
**Department/School:** Chemistry
**General Education and other Course Attributes:** Scientific Investigation, Natural Sciences

CHEM 1314 Chemistry I (LN)
**Prerequisites:** MATH 1483 or MATH 1513 or higher with a "C" or better or an acceptable math placement score (see placement.okstate.edu) or acceptable AP credit.

**Description:** The beginning chemistry course recommended for students in basic biological sciences (including pre-medical science and pre-veterinary sciences), physical sciences and engineering. This course covers chemical principles and their applications to the properties and transformations of matter, including periodic classification of the elements, laws of chemical combination, gas laws, atomic and molecular structure, and chemical bonding.

**Credit hours:** 4
**Contact hours:** Lecture: 3 Lab: 2 Contact: 5
**Levels:** Undergraduate
**Schedule types:** Lab, Lecture, Combined lecture and lab
**Department/School:** Chemistry
**General Education and other Course Attributes:** Scientific Investigation, Natural Sciences
CHEM 1414 General Chemistry for Engineers (LN)
Prerequisites: MATH 1483 or MATH 1513 or higher with a “C” or better or an acceptable math placement score (see placement.okstate.edu) or acceptable AP credit.
Description: One seminar survey of general chemistry for engineering students. Topics include physical properties of states of matter, stoichiometry, atomic theory, periodic properties, bonding, thermodynamics, equilibrium, acid-base and redox reactions, and elementary chemical thermodynamics. Topics will be discussed with respect to applications to materials, energy and environmental topics relevant to engineering students.
Credit hours: 4
Contact hours: Lecture: 3 Lab: 2 Contact: 5
Levels: Undergraduate
Schedule types: Lab, Lecture, Combined lecture and lab
Department/School: Chemistry
General Education and other Course Attributes: Scientific Investigation, Natural Sciences

CHEM 1515 Chemistry II (LN)
Prerequisites: CHEM 1314 with a grade of “C” or better or acceptable AP credit.
Description: A continuation of Chemistry 1 for students in the basic biological sciences (including premedical science and pre-veterinary science), physical sciences, and engineering. Topics include, but not limited to, intermolecular forces, liquids and solids, chemical equilibria, acid/base chemistry, oxidation/reduction, electrochemistry, chemical kinetics, and elementary chemical thermodynamics.
Credit hours: 5
Contact hours: Lecture: 3 Lab: 2 Contact: 6 Other: 1
Levels: Undergraduate
Schedule types: Discussion, Lab, Lecture, Combined lecture lab & disc
Department/School: Chemistry
General Education and other Course Attributes: Scientific Investigation, Natural Sciences

CHEM 2113 Principles of Analytical Chemistry
Prerequisites: A grade of “C” or better in CHEM 1515.
Description: Statistical analysis of analytical data, acid-base equilibria, acid-base titrations, electrochemistry, analytical separations, as well as atomic and molecular optical spectroscopy.
Credit hours: 3
Contact hours: Lecture: 3 Contact: 3
Levels: Undergraduate
Schedule types: Lecture
Department/School: Chemistry

CHEM 2122 Quantitative Analysis Laboratory
Prerequisites: CHEM 2113 or concurrent enrollment.
Description: Laboratory exercises related to theoretical principles in CHEM 2113.
Credit hours: 2
Contact hours: Lab: 4 Contact: 4
Levels: Undergraduate
Schedule types: Lab
Department/School: Chemistry

CHEM 2113 or concurrent enrollment.
Description: Laboratory exercises related to theoretical principles in CHEM 2113.
Credit hours: 2
Contact hours: Lab: 4 Contact: 4
Levels: Undergraduate
Schedule types: Lab
Department/School: Chemistry

CHEM 2890 Honors Experience in Chemistry
Prerequisites: Honors Program participation and concurrent enrollment in designated course(s).
Description: A supplemental Honors experience in Chemistry to partner concurrently with designated lower-division CHEM course(s). This course adds a different intellectual dimension to designated course(s).
Credit hours: 1
Contact hours: Lecture: 1 Contact: 1
Levels: Undergraduate
Schedule types: Lecture
Department/School: Chemistry
General Education and other Course Attributes: Honors Credit

CHEM 2980 Current Topics for Chemical Professionals
Prerequisites: Current enrollment in CHEM 1314 or higher chemistry course.
Description: Current topics for pre-chemical professionals which may include, but are not limited to; Chemistry of Life; Energy; Environmental; Materials; Energy; What’s that Stuff?, and Teaching/Learning. The course is intended to provide interested undergraduates with a broader introduction to topics relevant to future trends in chemistry and chemically-related fields. Discussion will be directed by faculty members with expertise in the identified area. Offered for variable credit, 1-6 credit hours, maximum of 6 credit hours.
Credit hours: 1-6
Contact hours: Lecture: 1-6 Contact: 1-6
Levels: Undergraduate
Schedule types: Lecture
Department/School: Chemistry

CHEM 2990 Special Problems in Chemistry
Prerequisites: CHEM 1314 or concurrent enrollment.
Description: Training in independent work, study of relevant literature and experimental investigation of an assigned problem at the lower-division level. Offered for variable credit, 1-3 credit hours, maximum of 3 credit hours.
Credit hours: 1-3
Contact hours: Lab: 2-6 Contact: 2-6
Levels: Undergraduate
Schedule types: Lab
Department/School: Chemistry

CHEM 3012 Survey of Organic Chemistry Laboratory
Prerequisites: CHEM 3013 or concurrent enrollment.
Description: Laboratory exercises related to theoretical principles covered in CHEM 3013.
Credit hours: 2
Contact hours: Lab: 3 Contact: 4 Other: 1
Levels: Undergraduate
Schedule types: Discussion, Lab, Combined Lab & Discussion
Department/School: Chemistry

CHEM 3013 Survey of Organic Chemistry
Prerequisites: A minimum grade of “C” in CHEM 1225 or CHEM 1515.
Description: Terminal, one-semester organic chemistry lecture course covering the general principles of nomenclature, structure, bonding, methods of preparation, reactions and use of acyclic, cyclic, and aromatic compounds.
Credit hours: 3
Contact hours: Lecture: 3 Contact: 3
Levels: Undergraduate
Schedule types: Lecture
Department/School: Chemistry
<table>
<thead>
<tr>
<th>Course Code</th>
<th>Course Name</th>
<th>Prerequisites</th>
<th>Description</th>
<th>Credit hours</th>
<th>Contact hours: Lecture</th>
<th>Contact</th>
<th>Levels</th>
<th>Schedule types</th>
<th>Department/School</th>
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<tbody>
<tr>
<td>CHEM 3015</td>
<td>Survey of Organic Chemistry</td>
<td>A grade of &quot;C&quot; or better in CHEM 1225 or CHEM 1515.</td>
<td>Terminal, one-semester course with a laboratory in organic chemistry covering the general principles of nomenclature, structure, bonding, methods of preparation, reactions and uses of acyclic, cyclic, and aromatic compounds. May not be used for degree credit with CHEM 3013.</td>
<td>5</td>
<td>Lecture: 3</td>
<td>Lab: 3</td>
<td>Lecture, Combined lecture and lab</td>
<td>Chemistry</td>
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<tr>
<td>CHEM 3053</td>
<td>Organic Chemistry I</td>
<td>A grade of &quot;C&quot; or better in CHEM 1515.</td>
<td>This course is the first of the in-depth sequence of organic chemistry. Topics include nomenclature, structure, stereochemistry, reactivity, properties, and synthesis of organic molecules with an emphasis on reaction mechanisms. This course is required for many life and physical science majors and pre-health students. Consult your degree requirements and professional school admission requirements.</td>
<td>3</td>
<td>Lecture: 3</td>
<td>Contact: 3</td>
<td>Undergraduate</td>
<td>Lecture</td>
<td>Chemistry</td>
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<tr>
<td>CHEM 3112</td>
<td>Organic Chemistry Laboratory</td>
<td>Completion of CHEM 3153 or concurrent enrollment.</td>
<td>Modern laboratory instrumentation, experimental techniques, and computational methods in physical chemistry.</td>
<td>2</td>
<td>Lab: 3</td>
<td>Contact: 4</td>
<td>Undergraduate</td>
<td>Discussion, Lab, Combined Lab &amp; Discussion</td>
<td>Chemistry</td>
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<tr>
<td>CHEM 3153</td>
<td>Organic Chemistry II</td>
<td>A grade of &quot;C&quot; or higher in CHEM 3053.</td>
<td>This course is the second of the in-depth sequence of organic chemistry starting with CHEM 3053.</td>
<td>3</td>
<td>Lecture: 3</td>
<td>Contact: 3</td>
<td>Graduate, Undergraduate</td>
<td>Lecture</td>
<td>Chemistry</td>
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<tr>
<td>CHEM 3353</td>
<td>Descriptive Inorganic Chemistry</td>
<td>A grade of &quot;C&quot; or higher in CHEM 1515, CHEM 1225.</td>
<td>Structures and properties of the elements and their many compounds in the broadest sense which includes the modern technologically important materials, organometallics, and inorganic substances of biological significance.</td>
<td>3</td>
<td>Lecture: 3</td>
<td>Contact: 3</td>
<td>Undergraduate</td>
<td>Lecture</td>
<td>Chemistry</td>
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<tr>
<td>CHEM 3363</td>
<td>Bioinorganic Chemistry</td>
<td>Grade of &quot;C&quot; or higher in CHEM 1225 or CHEM 1515 or acceptable AP credit.</td>
<td>Discusses the structural and functional roles of main group and transition metals within biological systems. Topics may include: the transport, distribution and properties of metals in biological systems, the coordination chemistry of biologically active metals, physical methods for determining metalloprotein structure and reactivity, chemical processes including redox processes and long-range electron transfer reactions and metallocofactors and metal clusters.</td>
<td>3</td>
<td>Lecture: 3</td>
<td>Contact: 3</td>
<td>Undergraduate</td>
<td>Lecture</td>
<td>Chemistry</td>
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<tr>
<td>CHEM 3413</td>
<td>Physical Chemistry Applications</td>
<td>Minimum grade of &quot;C&quot; or higher in both CHEM 1515 and MATH 2144.</td>
<td>A practical and applied approach to key topics in physical chemistry, including thermodynamics, chemical equilibria, and chemical kinetics, and how they relate to general chemical and biological processes on a molecular and macroscopic level.</td>
<td>3</td>
<td>Lecture: 3</td>
<td>Contact: 3</td>
<td>Undergraduate</td>
<td>Lecture</td>
<td>Chemistry</td>
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<tr>
<td>CHEM 3433</td>
<td>Physical Chemistry I</td>
<td>Minimum grade of &quot;C&quot; or higher in: CHEM 1515 and MATH 2153 and PHYS 2114.</td>
<td>Introductory theoretical analysis of molecular structure, chemical bonding and macroscopic chemical systems using quantum theory, classical and statistical thermodynamics, and kinetics. Previously offered as CHEM 3434.</td>
<td>3</td>
<td>Lecture: 3</td>
<td>Contact: 3</td>
<td>Undergraduate</td>
<td>Lecture</td>
<td>Chemistry</td>
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<tr>
<td>CHEM 3512</td>
<td>Physical Chemistry Laboratory</td>
<td>A grade of &quot;C&quot; or better in CHEM 2122 and CHEM 3433.</td>
<td>Modern laboratory instrumentation, experimental techniques, and computational methods in physical chemistry.</td>
<td>2</td>
<td>Lab: 3</td>
<td>Contact: 4</td>
<td>Undergraduate</td>
<td>Discussion, Lab, Combined Lab &amp; Discussion</td>
<td>Chemistry</td>
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<tr>
<td>CHEM 3553</td>
<td>Physical Chemistry II</td>
<td>A grade of &quot;C&quot; or better in CHEM 3433.</td>
<td>A continuation of CHEM 3433.</td>
<td>3</td>
<td>Lecture: 3</td>
<td>Contact: 3</td>
<td>Undergraduate</td>
<td>Lecture</td>
<td>Chemistry</td>
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Department/School: Chemistry
CHEM 3890 Advanced Honors Experience in Chemistry
Prerequisites: Honors Program participation and concurrent enrollment in designated course(s).
Description: A supplemental Honors experience in Chemistry to partner concurrently with designated upper-division CHEM course(s). This course adds a different intellectual dimension to designated course(s).
Credit hours: 1
Contact hours: Lecture: 1 Contact: 1
Levels: Undergraduate
Schedule types: Lecture
Department/School: Chemistry
General Education and other Course Attributes: Honors Credit

CHEM 4022 Modern Methods of Chemical Analysis Laboratory
Prerequisites: CHEM 4023 or concurrent enrollment.
Description: Laboratory exercises related to theoretical principles covered in CHEM 4023. May not be used for degree credit with CHEM 4020.
Credit hours: 2
Contact hours: Lab: 4 Contact: 4
Levels: Undergraduate
Schedule types: Lab
Department/School: Chemistry

CHEM 4023 Modern Methods of Chemical Analysis
Prerequisites: A grade of "C" or better in CHEM 2122; and CHEM 3413 or CHEM 3433.
Description: The design, operational principles and practical application of modern instrumental methods used in chemical analysis of natural and artificial materials. Covers the reagents and instruments used in the separation, identification and quantification of the chemical components. May not be used for degree credit with CHEM 4020.
Credit hours: 3
Contact hours: Lecture: 3 Contact: 3
Levels: Graduate, Undergraduate
Schedule types: Lecture
Department/School: Chemistry

CHEM 4123 Biomolecular Chemistry and Function
Prerequisites: Minimum grade of "C" in CHEM 3153, CHEM 3112, and CHEM 3413 or CHEM 3433.
Description: The class is designed to use examples from classic and current literature to expand the student's knowledge of the chemical techniques required to understand the structure and function of macromolecules in solution. These topics include chemical forces that stabilize macromolecular and supramolecular structure, thermodynamics and statistical mechanics of macromolecular and polymer folding, diffusional processes, kinetics, and the relationship of these principles to practical application in experimental design and interpretation.
Credit hours: 3
Contact hours: Lecture: 3 Contact: 3
Levels: Undergraduate
Schedule types: Lecture
Department/School: Chemistry

CHEM 4312 Inorganic Chemistry Laboratory
Prerequisites: Minimum grade of "C" or better in CHEM 3112.
Description: Course will provide students with practical knowledge and experimental techniques commonly used in inorganic and organometallic chemistry.
Credit hours: 2
Contact hours: Lab: 2 Contact: 3 Other: 1
Levels: Undergraduate
Schedule types: Discussion, Lab, Combined Lab & Discussion
Department/School: Chemistry

CHEM 4313 Medicinal Organic Chemistry
Prerequisites: Minimum grade of "C" in CHEM 3153 and CHEM 3112.
Description: This course looks at the development of new organic molecules for use in the pharmaceutical industry and investigates their pathway from the design stage to eventual introduction to the market. This course explores a range of important techniques necessary for the synthesis of complex organic architectures, an introduction to asymmetric synthesis, and polymer-supported synthesis of biomolecules including peptides and nucleic acids. This course also introduces various classes of drugs, mechanisms of action, drug metabolism and structure activity relationships.
Credit hours: 3
Contact hours: Lecture: 3 Contact: 3
Levels: Graduate, Undergraduate
Schedule types: Lecture
Department/School: Chemistry

CHEM 4320 Chemical and Spectrometric Identification of Organic Compounds
Prerequisites: A grade of "C" or higher in CHEM 3112 and CHEM 3153.
Description: Theory and practice in separating mixtures of organic compounds and some theory and practice in identifying organic compounds by spectroscopic methods. Offered for variable credit, 1-3 credit hours, maximum of 3 credit hours.
Credit hours: 1-3
Contact hours: Lab: 2-6 Contact: 2-6
Levels: Graduate, Undergraduate
Schedule types: Lab
Department/School: Chemistry

CHEM 4333 Inorganic Chemistry I
Prerequisites: CHEM 1515 with minimum grade of "C."
Description: Bonding theory, molecular symmetry and its applications to structure, bonding and spectroscopy, structures of simple solids, inorganic acids and bases, oxidation and reduction, and industrial production of elements, coordination chemistry, crystal field theory, ligand field theory, introduction to organometallic chemistry. May not be used for degree credit with CHEM 5260.
Credit hours: 3
Contact hours: Lecture: 3 Contact: 3
Levels: Undergraduate
Schedule types: Lecture
Department/School: Chemistry

CHEM 4990 Special Problems in Chemistry
Prerequisites: Instructor permission required.
Description: Training in independent work, study of relevant literature and experimental investigation of an assigned problem culminating in a written and oral report. Offered for variable credit, 1-6 credit hours, maximum of 6 credit hours.
Credit hours: 1-6
Contact hours: Contact: 1-6 Other: 1-6
Levels: Graduate, Undergraduate
Schedule types: Independent Study
Department/School: Chemistry
CHEM 5000 Thesis
Description: Familiarizes the student with methods used in research in chemistry. Offered for variable credit, 1-6 credit hours, maximum of 6 credit hours.
Credit hours: 1-6
Contact hours: Contact: 1 Other: 1-6
Levels: Graduate
Schedule types: Independent Study
Department/School: Chemistry

CHEM 5001 Introduction to Chemistry Research
Prerequisites: Graduate standing.
Description: Introduction to chemical research topics of interest to the department. Special emphasis placed on ethics, plagiarism, codes of conduct, research notebooks, publishing, and presentations.
Credit hours: 1
Contact hours: Lecture: 1 Contact: 1
Levels: Graduate
Schedule types: Lecture
Department/School: Chemistry

CHEM 5011 Graduate Seminar
Description: Preparation and presentation of seminars usually on subjects of current interest taken from the literature. Completion of 1 credit hour required for MS degree.
Credit hours: 1
Contact hours: Contact: 1 Other: 1
Levels: Graduate
Schedule types: Discussion
Department/School: Chemistry

CHEM 5053 Foundations of Physical Chemistry
Prerequisites: BS/BA in chemistry, CHEM 3153, or equivalent.
Description: This course provides the foundations of physical chemistry required for all disciplines of chemistry to understand the underlying principles necessary to advance at the graduate level. This core treatment will address thermodynamics and equilibria, chemical kinetics, quantum mechanics, spectroscopy, and statistical thermodynamics. These topics will provide the conceptual learning critical for interdisciplinary applications of physical chemistry.
Credit hours: 3
Contact hours: Lecture: 3 Contact: 3
Levels: Graduate
Schedule types: Lecture
Department/School: Chemistry

CHEM 5057 Foundations of Analytical Chemistry
Prerequisites: BS/BA in chemistry, CHEM 3153, or equivalent.
Description: This course provides the basic principles of analytical chemistry necessary to advance at the graduate level in all disciplines of chemistry. Subject matter includes the underlying principles of chemical analyses with emphasis on chemical and biological reactions (equilibrium, reaction rate, chemical labeling), instrumentation and instrumental design, sampling, sample preparation and method validation. These topics will provide the conceptual foundation critical for interdisciplinary applications of analytical chemistry.
Credit hours: 3
Contact hours: Lecture: 3 Contact: 3
Levels: Graduate
Schedule types: Lecture
Department/School: Chemistry

CHEM 5073 Foundations of Analytical Chemistry
Prerequisites: One year of physical chemistry.
Description: Principles of bulk and multi-stage separation methods: chromatography, liquid-liquid extraction, and zone melting.
Credit hours: 3
Contact hours: Lecture: 3 Contact: 3
Levels: Graduate
Schedule types: Lecture
Department/School: Chemistry

CHEM 5103 Physical and Chemical Separations
Prerequisites: One year of physical chemistry.
Description: Principles of bulk and multi-stage separation methods: chromatography, liquid-liquid extraction, and zone melting.
Credit hours: 3
Contact hours: Lecture: 3 Contact: 3
Levels: Graduate
Schedule types: Lecture
Department/School: Chemistry

CHEM 5113 Equilibrium and Kinetics in Analytical Chemistry
Prerequisites: One year of physical chemistry.
Description: Physical and chemical principles of equilibrium and kinetics as applied to analytical problems.
Credit hours: 3
Contact hours: Lecture: 3 Contact: 3
Levels: Graduate
Schedule types: Lecture
Department/School: Chemistry

CHEM 5211 Equilibrium and Kinetics in Organic Chemistry
Prerequisites: CHEM 3153 and CHEM 3433 or equivalent.
Description: Principles of bulk and multi-stage separation methods: chromatography, liquid-liquid extraction, and zone melting.
Credit hours: 3
Contact hours: Lecture: 3 Contact: 3
Levels: Graduate
Schedule types: Lecture
Department/School: Chemistry

CHEM 5220 Topics For Teachers
Prerequisites: Teaching experience.
Description: Designed to help elementary and secondary science teachers improve their subject matter competence in chemistry. Content varies depending on the needs of specific groups of teachers. Offered for variable credit, 1-9 credit hours, maximum of 9 credit hours.
Credit hours: 1-9
Contact hours: Contact: 1-9 Other: 1-9
Levels: Graduate
Schedule types: Independent Study
Department/School: Chemistry

CHEM 5223 Polymer Chemistry
Prerequisites: CHEM 3153 and CHEM 3433 or equivalent.
Description: Preparation and polymerization of organic monomers; properties and uses of resulting high polymers; theories of polymerization; inorganic and natural organic polymers.
Credit hours: 3
Contact hours: Lecture: 3 Contact: 3
Levels: Graduate
Schedule types: Lecture
Department/School: Chemistry
CHEM 5263 Foundations of Inorganic Chemistry  
**Prerequisites:** CHEM 1515 with minimum grade of “C.”  
**Description:** Bonding theory, molecular symmetry and its applications to structure, bonding and spectroscopy, structures of simple solids, inorganic acids and bases, oxidation and reduction, and industrial production of elements, coordination chemistry, crystal field theory, ligand field theory, introduction to organometallic chemistry. May not be used for degree credit with CHEM 4333. Previously offered as CHEM 5260.  
**Credit hours:** 3  
**Contact hours:** Lecture: 3  
**Levels:** Graduate  
**Schedule types:** Lecture  
**Department/School:** Chemistry  

CHEM 5283 Solid State Chemistry  
**Prerequisites:** CHEM 5260.  
**Description:** Structure, bonding, and properties of crystalline and amorphous inorganic solids. Emphasis on the characterization of inorganic solids and phase transitions in inorganic solids.  
**Credit hours:** 3  
**Contact hours:** Lecture: 3  
**Levels:** Graduate  
**Schedule types:** Lecture  
**Department/School:** Chemistry  

CHEM 5323 Reactions of Organic Compounds  
**Prerequisites:** CHEM 3153.  
**Description:** Products and mechanisms of reactions of importance in organic synthesis.  
**Credit hours:** 3  
**Contact hours:** Lecture: 3  
**Levels:** Graduate  
**Schedule types:** Lecture  
**Department/School:** Chemistry  

CHEM 5373 Spectrometric Identification of Organic Compounds  
**Prerequisites:** CHEM 4320.  
**Description:** Lectures on ultraviolet, circular dichroism, infrared, nuclear magnetic resonance (NMR) and mass spectrometry (MS). More advanced techniques in NMR and MS stressed. Hands-on training and use of modern spectroscopic instrumentation in laboratory.  
**Credit hours:** 3  
**Contact hours:** Lecture: 3  
**Levels:** Graduate  
**Schedule types:** Lecture  
**Department/School:** Chemistry  

CHEM 5433 Computational Chemistry and Molecular Modeling  
**Credit hours:** 3  
**Contact hours:** Lecture: 3  
**Levels:** Graduate  
**Schedule types:** Lecture  
**Department/School:** Chemistry  

CHEM 5443 Mechanism and Structure in Organic Chemistry  
**Prerequisites:** CHEM 3153 and CHEM 3553.  
**Description:** Relationship of properties of organic compounds to their structure; mechanisms of organic reactions.  
**Credit hours:** 3  
**Contact hours:** Lecture: 3  
**Levels:** Graduate  
**Schedule types:** Lecture  
**Department/School:** Chemistry  

CHEM 5563 Chemical Thermodynamics I  
**Prerequisites:** CHEM 3553.  
**Description:** Statistical and classical thermodynamics applied to chemical systems.  
**Credit hours:** 3  
**Contact hours:** Lecture: 3  
**Levels:** Graduate  
**Schedule types:** Lecture  
**Department/School:** Chemistry  

CHEM 5623 Quantum Chemistry I  
**Prerequisites:** CHEM 3553.  
**Description:** Fundamentals of quantum mechanics, including classical mechanics, wave representation of matter, the Schroedinger equation, and atomic structure.  
**Credit hours:** 3  
**Contact hours:** Lecture: 3  
**Levels:** Graduate  
**Schedule types:** Lecture  
**Department/School:** Chemistry  

CHEM 5963 Advanced Inorganic Chemistry  
**Prerequisites:** CHEM 5263.  
**Description:** Inorganic reaction mechanisms, catalysis, electronic spectra of complexes, luminescence of inorganic compounds, lanthanide and actinide chemistry, introduction to biological inorganic chemistry. Previously offered as CHEM 5960.  
**Credit hours:** 3  
**Contact hours:** Lecture: 3  
**Levels:** Graduate  
**Schedule types:** Lecture  
**Department/School:** Chemistry  

CHEM 6000 Doctoral Dissertation Research  
**Prerequisites:** MS degree in chemistry or consent of instructor.  
**Description:** Independent investigation under the direction and supervision of a major professor. Offered for variable credit, 1-15 credit hours, maximum of 60 credit hours.  
**Credit hours:** 1-15  
**Contact hours:** Contact: 1-15  
**Levels:** Graduate  
**Schedule types:** Independent Study  
**Department/School:** Chemistry  

CHEM 6010 Research Seminar  
**Prerequisites:** Consent of instructor.  
**Description:** Participation in departmental seminars on current topics in chemistry. One credit hour each fall and spring required for MS and PhD candidates with the exception of the first semester. Offered for variable credit, 1-20 credit hours, maximum of 20 credit hours.  
**Credit hours:** 1-20  
**Contact hours:** Contact: 1-20  
**Levels:** Graduate  
**Schedule types:** Independent Study  
**Department/School:** Chemistry
CHEM 6011 Advanced Seminar  
**Prerequisites:** CHEM 5011 or MS degree.  
**Description:** Preparation and oral presentation of critical reviews on chemical subjects. Usually related to the student’s research area. Completion of one credit hour required for the PhD degree.  
**Credit hours:** 1  
**Contact hours:** Contact: 1 Other: 1  
**Levels:** Graduate  
**Schedule types:** Discussion  
**Department/School:** Chemistry

CHEM 6050 Special Topics in Analytical Chemistry  
**Description:** Supervised study of topics and fields not otherwise covered. Offered for variable credit, 1-6 credit hours, maximum of 6 credit hours.  
**Credit hours:** 1-6  
**Contact hours:** Contact: 1-6 Other: 1-6  
**Levels:** Graduate  
**Schedule types:** Independent Study  
**Department/School:** Chemistry

CHEM 6103 Electroanalytical Chemistry  
**Prerequisites:** CHEM 4024.  
**Description:** The theory, practice and instrumentation in various areas of modern electroanalytical chemistry.  
**Credit hours:** 3  
**Contact hours:** Lecture: 3 Contact: 3  
**Levels:** Graduate  
**Schedule types:** Lecture  
**Department/School:** Chemistry

CHEM 6113 Analytical Spectroscopy  
**Prerequisites:** CHEM 4024.  
**Description:** Survey of selected topics in analytical applications of spectroscopic techniques. Fundamental concepts as well as current trends in research, including instrumentation.  
**Credit hours:** 3  
**Contact hours:** Lecture: 3 Contact: 3  
**Levels:** Graduate  
**Schedule types:** Lecture  
**Department/School:** Chemistry

CHEM 6223 Physical Polymer Science  
**Prerequisites:** CHEM 5223 or equivalent.  
**Description:** A study of the physical properties of macromolecular systems including polymer solutions, gels, bulk polymers and rubbers. The characterization of polymers based on their thermal, spectroscopic, microstructure and molecular masses is also discussed.  
**Credit hours:** 3  
**Contact hours:** Lecture: 3 Contact: 3  
**Levels:** Graduate  
**Schedule types:** Lecture  
**Department/School:** Chemistry

CHEM 6303 Physical Organic Chemistry  
**Prerequisites:** BS/BA in chemistry, CHEM 3153, or equivalent.  
**Description:** This course is an examination of the methods used in organic chemistry to probe mechanisms and reactive intermediates. Topics will include isotope effects, kinetics, linear free energy relationships, an introduction of orbital symmetry, rearrangements, stereo electronic effects, the generation and chemistry of carbenium ions, carbanions, carbenes, radicals, excited states, and strained molecules.  
**Credit hours:** 3  
**Contact hours:** Lecture: 3 Contact: 3  
**Levels:** Graduate  
**Schedule types:** Lecture  
**Department/School:** Chemistry

CHEM 6420 Special Topics in Organic Chemistry  
**Prerequisites:** CHEM 3153.  
**Description:** Deals with topics not covered in other courses. Offered for variable credit, 1-9 credit hours, maximum of 9 credit hours.  
**Credit hours:** 1-9  
**Contact hours:** Contact: 1-9 Other: 1-9  
**Levels:** Graduate  
**Schedule types:** Independent Study  
**Department/School:** Chemistry

CHEM 6453 Chemical Kinetics  
**Prerequisites:** CHEM 3553.  
**Description:** The kinetics of chemical reactions and their theoretical interpretation.  
**Credit hours:** 3  
**Contact hours:** Lecture: 3 Contact: 3  
**Levels:** Graduate  
**Schedule types:** Lecture  
**Department/School:** Chemistry

CHEM 6553 Molecular Spectroscopy  
**Prerequisites:** CHEM 5623.  
**Description:** Spectra and structure of molecules.  
**Credit hours:** 3  
**Contact hours:** Lecture: 3 Contact: 3  
**Levels:** Graduate  
**Schedule types:** Lecture  
**Department/School:** Chemistry

CHEM 6650 Selected Topics in Advanced Physical and Inorganic Chemistry  
**Prerequisites:** Consent of instructor.  
**Description:** Supervised study of selected topics and fields not otherwise covered. Same course as CHEM 5960. Offered for variable credit, 1-6 credit hours, maximum of 12 credit hours.  
**Credit hours:** 1-6  
**Contact hours:** Contact: 1-6 Other: 1-6  
**Levels:** Graduate  
**Schedule types:** Independent Study  
**Department/School:** Chemistry
CHEM 6803 Photonics I: Advanced Optics  
**Prerequisites:** ECEN 3813 or PHYS 3213, or consent of instructor.  
**Description:** Advanced optics, including spectral and time characteristics of detectors, characteristics of lasers, time, spectral and spatial parameters of laser emission, interferometric techniques, and nonlinear effects such as two-photon absorption and second and third harmonic generations. Ultra short laser pulses. Same course as ECEN 6803 & PHYS 6803. Offered for fixed credit, maximum of 9 credit hours.  
**Credit hours:** 3  
**Contact hours:** Lab: 6 Contact: 6  
**Levels:** Graduate  
**Schedule types:** Lab  
**Department/School:** Chemistry

CHEM 6810 Photonics II: THz photonics and THz time-domain spectroscopy  
**Prerequisites:** CHEM 6803.  
**Description:** THz photonics and THz time-domain spectroscopy (THz-TDS). Concepts and techniques of driving electronic circuitry with ultra short laser pulses to generate and detect freely propagating pulses of THz electromagnetic radiation using several operational research systems. Same course as ECEN 6810 & PHYS 6810. Offered for fixed credit, maximum of 4 credit hours.  
**Credit hours:** 1  
**Contact hours:** Lab: 2 Contact: 2  
**Levels:** Graduate  
**Schedule types:** Lab  
**Department/School:** Chemistry

CHEM 6811 Photonics II: THz Photonics and THz - TDS  
**Prerequisites:** CHEM 6803.  
**Description:** THz photonics and THz time-domain spectroscopy (THz-TDS). Concepts and techniques of driving electronic circuitry with ultrashort laser pulses to generate and detect freely propagating pulses of THz electromagnetic radiation using several operational research systems. Same course as ECEN 6811 and PHYS 6811.  
**Credit hours:** 1  
**Contact hours:** Lecture: 1 Contact: 3  
**Levels:** Graduate  
**Schedule types:** Lecture  
**Department/School:** Chemistry

CHEM 6820 Photonics II: Spectroscopy II  
**Prerequisites:** CHEM 6803.  
**Description:** Operating principles and applications of laser spectroscopy of atoms, molecules, solids and complex fluids. Absorption, emission, photon correlation, coherence, time resolved Fourier transform. Raman spectroscopy and non-linear optical. Same course as ECEN 6820 & PHYS 6820. Offered for fixed credit, 1 credit hour(s), maximum of 4 credit hours.  
**Credit hours:** 1  
**Contact hours:** Lab: 2 Contact: 2  
**Levels:** Graduate  
**Schedule types:** Lab  
**Department/School:** Chemistry

CHEM 6830 Photonics II: Spectroscopy III  
**Prerequisites:** CHEM 6803.  
**Description:** Advanced spectroscopic instruments and methods used for investigation of semi-conductors and solid state material. Stimulated emission characterized both in wavelength and in time. Time-resolved fluorescence measurements. Multiphotonic excitations. Fast measuring techniques, including subnanosecond detectors, picosecond streak cameras, and ultra fast four-wave mixing and correlation techniques.  
**Credit hours:** 1  
**Contact hours:** Lab: 2 Contact: 2  
**Levels:** Graduate  
**Schedule types:** Lab  
**Department/School:** Chemistry

CHEM 6840 Photonics III: Microscopy II  
**Prerequisites:** CHEM 3553 or consent of instructor.  
**Description:** The structure and imaging of solid surfaces. Basics of scanning probe microscopy (SPM). Contact and noncontact atomic force microscopy (AFM). Scanning tunneling microscopy (STM) in air.  
**Credit hours:** 1  
**Contact hours:** Lab: 2 Contact: 2  
**Levels:** Graduate  
**Schedule types:** Lab  
**Department/School:** Chemistry

CHEM 6850 Photonics III: Microscopy II  
**Prerequisites:** CHEM 3553 or consent of instructor.  
**Description:** Advanced techniques of scanning probe microscopy (SPM). Magnetic force microscopy, Kelvin force microscopy, scanning tunneling microscopy (STM) in vacuum. Characterization of materials with SPM. Nanolithography with SPM. Device manufacturing and analysis. Same course as ECEN 6850 & PHYS 6850. Offered for fixed credit, 1 credit hour(s), maximum of 4 credit hours.  
**Credit hours:** 1  
**Contact hours:** Lab: 2 Contact: 2  
**Levels:** Graduate  
**Schedule types:** Lab  
**Department/School:** Chemistry

CHEM 6860 Photonics III: Microscopy III and Image Processing  
**Prerequisites:** ECEN 5793.  
**Description:** Digital image processing, including projects. Image acquisition and display, image enhancement, geometric operations, linear and nonlinear filtering, image restoration, edge detection, image analysis, morphology, segmentation, recognition, and coding/compression. Same course as ECEN 6860 & PHYS 6860. Offered for fixed credit, 1 credit hour(s), maximum of 4 credit hours.  
**Credit hours:** 1  
**Contact hours:** Lab: 2 Contact: 2  
**Levels:** Graduate  
**Schedule types:** Lab  
**Department/School:** Chemistry
CHEM 6870 Photonics IV: Synthesis and Devices I
Prerequisites: CHEM 6803 and CHEM 6840.
Description: Preparation of functional nanostructures and related optical and electronic devices. Physical and chemical methods of thin film deposition. Engineering of prototypes of light emitting diodes, sensors, optical limiting coatings, lithographic patterns. Same course as ECEN 6870 & PHY 6870. Offered for variable credit, 1 credit hour(s), maximum of 4 credit hours.
Credit hours: 1
Contact hours: Lab: 2 Contact: 2
Levels: Graduate
Schedule types: Lab
Department/School: Chemistry

CHEM 6880 Photonics IV: Semiconductor Devices, Testing and Characterization
Prerequisites: CHEM 6803.
Description: Test and characterization of semiconductor and optoelectronic devices. Hall Effect, four point probe, CV and IV measurements, optical pump-probe, photoluminescence and electro-optics sampling. Same course as ECEN 6880 & PHY 6880. Offered for fixed credit, 1 credit hour(s), maximum of 4 credit hours.
Credit hours: 1
Contact hours: Lab: 2 Contact: 2
Levels: Graduate
Schedule types: Lab
Department/School: Chemistry

CHEM 6890 Photonics IV: Semiconductor Synthesis and Devices III
Prerequisites: CHEM 6803.
Description: Processing, fabrication and characterization of semiconductor optoelectronic devices in class 100/10000 clean rooms. Clean room operation, including general procedure for material processing and device fabrication. Device processing using a variety of processing such as mask aligner, vacuum evaporators and rapid thermal annealer. Testing using optical and electrical testing apparatus such as I-V, C-V, Hall and optical spectral measurement systems. Same course as ECEN 6890 & PHY 6890. Offered for variable credit, 1 credit hour(s), maximum of 4 credit hours.
Credit hours: 1
Contact hours: Lab: 2 Contact: 2
Levels: Graduate
Schedule types: Lab
Department/School: Chemistry

Undergraduate Programs
- Biochemistry, BS (http://catalog.okstate.edu/arts-sciences/chemistry/biochemistry-bs)
- Chemistry (Approved by the American Chemical Society), BS (http://catalog.okstate.edu/arts-sciences/chemistry/chemistry-acscs-bs)
- Chemistry: Departmental Degree, BS (http://catalog.okstate.edu/arts-sciences/chemistry/chemistry-departmental-degree-bs)
- Chemistry: Pre-Health/Pre-Law, BS (http://catalog.okstate.edu/arts-sciences/chemistry/chemistry-pre-health-pre-law-bs)
- Chemistry: Secondary Teacher Certification, BS (http://catalog.okstate.edu/arts-sciences/chemistry/chemistry-secondary-teacher-certification-bs)
- Medicinal and Biophysical Chemistry, BS (http://catalog.okstate.edu/arts-sciences/chemistry/medicinal-and-biophysical-chemistry-bs)
- Biochemistry (BIOC), Minor (http://catalog.okstate.edu/arts-sciences/chemistry/biochemistry-minor)
- Chemistry (CHEM), Minor (http://catalog.okstate.edu/arts-sciences/chemistry/chemistry-minor)

Graduate Programs
Prerequisites
Students entering this program should have at least eight semester credit hours (or the equivalent) in general, analytical, organic and physical chemistry. The physical chemistry should have been based on mathematics through calculus.

Admission Requirements
For admission, a grade-point average of 3.00 or better is generally required. Deserving applicants with grade-point averages less than 3.00 are occasionally admitted under probationary conditions. Additional support of the application is sought in the form of three letters of recommendation. Graduate Record Examination scores are not required. Recommendations on admission to the Graduate College are made on behalf of the applicant by the departmental admission officer. Acceptance by a permanent adviser is not a prerequisite for admission to the program.

Degree Requirements
A more detailed description of the graduate study program in chemistry is available in a brochure supplied by the department upon request, or on the Internet at http://chemistry.okstate.edu. The requirements set forth below complement the general requirements stated in the "Graduate College (http://catalog.okstate.edu/graduate-college)" section of the OSU Catalog. Attendance and participation in the departmental colloquia are required.

The Master of Science Degree
Students must complete at least 30 credit hours of graduate coursework in chemistry or related fields. Each student must present an acceptable thesis dealing with a research problem and pass a final oral examination covering it and related material. Research on the thesis problem should be started as early as possible in the graduate program.

The Doctor of Philosophy Degree
Work is offered which leads to the degree with a focus in analytical, biological, environmental, inorganic, materials, medicinal, nanotechnology, organic physical, polymer or theoretical chemistry or chemical education. The student must pass a qualifying examination in his or her field of specialization. An acceptable dissertation must be presented which contains a substantial original contribution to the field of chemistry. The student must pass a final oral examination covering the dissertation and related material. The Doctor of Philosophy degree requires the completion of at least 90 semester credit hours of work beyond the bachelor’s degree. The course requirements are determined by the student and his/her advisory committee consistent with departmental requirements.

Faculty
Nicholas Materer, PhD—Professor and Chair
Regents Professors: K. Darrell Berlin, PhD; Frank D. Blum, PhD (Harrison i. Bartlett Chair); Warren T. Ford, PhD (emeritus); Lionel M. Raff, PhD (emeritus); Ziad El Rassi, PhD
Professors: Allen W. Apblett, PhD; Richard A. Bunce, PhD; J. Paul Devlin, PhD (emeritus); John I. Gelder, PhD; Barry Lavine, PhD; Mark G. Rockley, PhD (emeritus); Sheryl Tucker, PhD; Charles S. Weinert, PhD
Associate Professors: Smita Mohanty, PhD
Assistant Professors: Jeannie Bolliger, PhD; Gabriel Cook, PhD; Christopher Fennell, PhD; Sadagopan Krishnan, PhD; Toby Nelson, PhD; Laleh Tahsini, PhD; Yolanda Vasquez, PhD; Jimmie Weaver, PhD